

## **Method Abstract #48**

# Fluoride by Standard Addition

### **Scope and Application**

This method determines the concentration of fluoride ions in a sample and helps overcome sample matrix difficulties.

### **Method Summary**

The concentration of fluoride is determined by a fluoride ion selective electrode (ISE). An initial reading is taken, following which a known amount of a higher concentration standard is added to the sample, via buret. A second measurement is then taken and based on the millivolt (mV) difference the concentration of the initial sample can be calculated. Both the standards and the samples are pre-treated with TISAB (Total Ionic Strength Adjuster) to ensure that the background response from each sample is similar.

# Sample Calibration Curve 200.0 100.0 100.0 1.00(10 -2) 1.00(10 -1) 1.00(10 0) 1.00(10 1) 1.00(10 2) 1.00(10 3

log(Concentration)

### **Method Performance**

Parameter	Specification
Measuring Range*	0.1 – 1000ppm
MDL**	0.1ppm
RSD @ 0.1ppm	3.60%
RSD @ 1ppm	8.75%
RSD @ 10ppm	1.61%
RSD @ 100ppm	2.23%

<sup>\*</sup>Data for this measuring range was obtained using laboratory prepared standards formulated from sodium fluoride. The measuring range may be increased by using auto-dilution.

<sup>\*\*</sup>The Method Detection Limit (MDL) was determined based on data obtaining a coefficient of variance better than 30%. Results may differ depending on laboratory practices and sample matrix.