

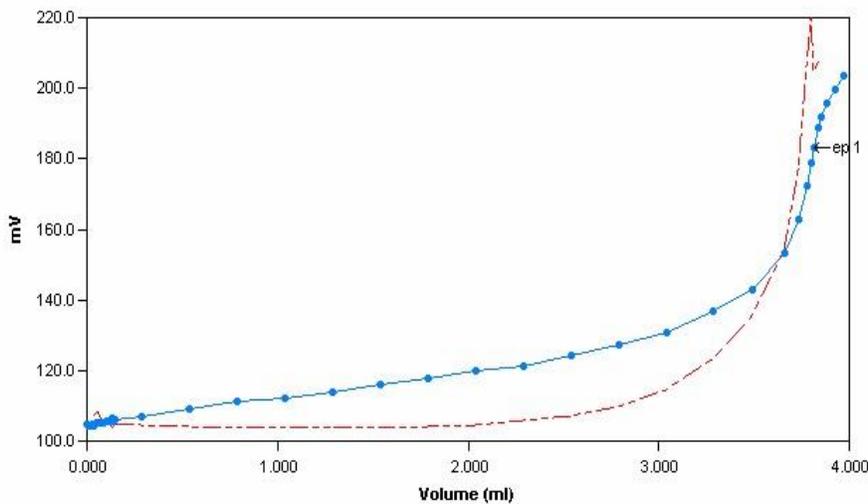
## Method Abstract #71

### Chloride by Titration

**Scope and Application** This method conforms to Standard Method 4500-Cl<sup>-</sup> D, and is a variation of ASTM D 512 (B) and ISO 9297. This method determines the chloride concentration of water samples.

**Method Summary** Chloride concentration is determined potentiometrically by titration with silver nitrate using a silver/sulfide ion selective electrode. The reaction involves the precipitation of silver chloride when chloride anions are combined with silver cations. The titration is complete when all the ions are consumed, signified by a large inflection.

**Sample Titration Curve**



**Method Performance**

Parameter	Specification
Measuring Range*	1 – 10,000ppm
MDL**	1ppm
RSD @ 1ppm	8.26% or +/- 0.08ppm
RSD @ 10ppm	0.95% or +/- 0.10ppm
RSD @ 100ppm	0.24% or +/- 0.24ppm
RSD @ 1000ppm	0.19% or +/- 1.90ppm

\*Data for this measuring range was obtained using laboratory prepared standards formulated from potassium chloride. Auto-Dilution or large analysis vessels may be required for analysis at upper end of measuring range.

\*\*The Method Detection Limit (MDL) was determined based on data obtaining a coefficient of variance better than 30%. Results may differ depending on laboratory practices and sample matrix.

RSD values are better than those specified in Standard Methods.